# Crystal Structure of $2 \alpha, 3 \alpha$-Epithio- $5 \alpha$-androstan-17 $\beta$-ol (R)-S-Oxide 

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The crystal structure of the title compound (I) has been determined by three-dimensional $X$-ray analysis from diffractometer data. Crystals are othorhombic, $a=22.806(5), b=6.276(2), c=11.978(3) \AA, Z=4$, space group $P 2_{1} 2_{1} 2_{1}$. The structure was solved by the heavy atom method and refined by full-matrix least-squares calculations to $R$ of 0.057 for 2065 independently observed reflections. Hydrogen atoms were located from a difference-Fourier synthesis. All the ring junctions are trans,trans. Rings B and C of the steroid skeleton have regular chair and rings $A$ and $D$ have distorted half-chair conformation.

As part of a continuing investigation into the structureactivity relationship in thiosteroid derivatives, ${ }^{1,2}$ the synthesis and reactions of $2 \alpha, 3 \alpha$-epithio- $5 \alpha$-androstan$17 \beta$-ol $(R)$-S-oxide (1) were recently investigated. ${ }^{3,4} \mathrm{We}$

(1)
have previously determined the structures of $2 \alpha, 3 \alpha-$ epithio- $5 \alpha$-androstan-17 $\beta$-yl $p$-bromobenzoate ${ }^{5}$ and $2 \alpha, 3 \alpha$-epithio- $5 \alpha$-androst- 6 -en-17 $\beta$-yl $p$-bromobenzoate, ${ }^{6}$ and found that introduction of the thiiran ring has some influence on the conformation of ring $A$. The present crystal structure determination was carried out in order to elucidate the detailed stereochemistry of (1),

[^0]particularly the direction of the $\mathrm{S}-\mathrm{O}$ bond at the sulphur atom in the thiiran ring.

## EXPERIMENTAL

Crystals of (1) were prepared ${ }^{7}$ and recrystallized from methanol as colourless prisms, m.p. $159-161^{\circ} \mathrm{C}$, elongated along the $b$ axis. Preliminary space-group data were determined from Weissenberg and precession photographs by use of $\mathrm{Cu}-K_{\alpha}(\lambda=1.5418 \AA)$ radiation, and accurate unit-cell dimensions were obtained by least-squares refinement of the setting angles of 20 reflections measured by automatic diffractometer.

Crystal Data. $-\mathrm{C}_{19} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{~S}, \quad M=322.5$. Orthorhombic, $a=22.806(5), b=6.276(2), c=11.978(3) \AA, U=1714 \AA^{3}$, $D_{\mathrm{m}}=1.244$ (by flotation), $Z=4, D_{\mathrm{c}}=1.249 \mathrm{~g} \mathrm{~cm}^{-3}$, $F(000)=704$. Space group $P 2_{1} 2_{1} 2_{1},\left(D_{2}{ }^{4}\right.$, No. 19), from systematic absences: $h 00$ for $h$ odd, $0 k 0$ for $k$ odd, and $00 l$ for $l$ odd. Mo- $K_{\alpha}$ radiation, $\lambda=0.7107 \AA ; \mu\left(\mathrm{Mo}-K_{\alpha}\right)=$ $1.933 \mathrm{~cm}^{-1}$.
Three-dimensional intensity data were collected on a Hilger and Watts automatic four-circle Y 290 diffractometer controlled by a PDP 8 computer. Integrated
${ }^{4}$ M. Kishi, S. Ishihara, and T. Komeno, Tetrahedron, 1974, 30, 2135.
${ }_{5}^{5}$ K. Utsumi-Oda and H. Koyama, J.C.S. Perkin II, 1973, 1866.
${ }^{6}$ K. Utsumi-Oda and H. Koyama, J.C.S. Perkin II, 1975, 993.
${ }^{7}$ S. Ishihara, ref. 5.
intensities were measured for $\theta<27.5^{\circ}$ by the $\theta-2 \theta$ scan technique with Mo- $K_{\alpha}$ radiation and a scintillation counter (with a Zirconium filter and pulse-height analyser). Each reflection was integrated in 80 steps of $0.01^{\circ}$, and a standard reflection monitored every 10 reflections. Scan counts ( $P$ ) and background counts ( $B_{1}, B_{2}$ ) were combined to yield the integrated intensities $I=P-2\left(B_{1}+B_{2}\right)$.

A total of 2286 independent reflections were recorded, of which 2065 having $I>3 \sigma(I)$ were considered observed. All intensities were corrected for Lorentz and polarization factors, and structure amplitudes were derived. No absorption corrections were applied since the specimen was considered to be sufficiently small $(0.55 \times 0.50 \times 0.55 \mathrm{~mm})$.

Structure Analysis.-The structure was solved by the SEARCHER programme for automatic heavy-atom analysis, written ${ }^{8,9}$ for the CDC 3600 computer (later modified for CDC 6600 computer). Initial co-ordinates for the sulphur atom were easily derived from a threedimensional Patterson synthesis. The atomic co-ordinates of 21 light atoms, which were treated as carbon atoms, were found from the first cycle of the SEARCHER programme (CDC 6600 computer).

The co-ordinates of the 22 atoms from the first SEARCHER cycle were refined ${ }^{10}$ by full-matrix leastsquares calculations. After three cycles, with isotropic

Table 1
Final fractional atomic co-ordinates ( $\times 10^{4}$ ), with estimated standard deviations in parentheses

| Atom | $x$ | $y$ | $z$ |
| :---: | :---: | :---: | :---: |
| S | 4 546(0) | -546(2) | 6 583(1) |
| O(1) | $5165(1)$ | -1098(6) | $6258(3)$ |
| $\mathrm{O}(2)$ | 73(1) | -848(6) | $5409(3)$ |
| C(1) | 3 451(1) | -528(8) | 5 387(4) |
| C(2) | 4 104(1) | 24(7) | $5335(3)$ |
| C(3) | $4324(1)$ | 1953 (8) | $5915(4)$ |
| C(4) | 3 913(1) | 3 451(8) | 6 489(5) |
| C(5) | 3 297(1) | 2 528(7) | 6 688(3) |
| C(6) | 2 893(2) | 4242 (8) | $7123(4)$ |
| C(7) | 2 275(1) | 3 448(8) | 7 350(4) |
| C(8) | $2012(1)$ | 2 366(6) | 6336 (3) |
| C(9) | 2 424(1) | 557 (7) | $5928(3)$ |
| $\mathrm{C}(10)$ | $3051(1)$ | $1387(6)$ | 5 639(3) |
| C(11) | 2 143(1) | -797(8) | 5001 (4) |
| C(12) | $1522(1)$ | -1 566(8) | 5 267(4) |
| $\mathrm{C}(13)$ | 1130 (1) | 277(7) | $5594(3)$ |
| C(14) | 1410 (1) | $1414(6)$ | 6 598(3) |
| C(15) | 922 (2) | 2 845(8) | 7 058(4) |
| $\mathrm{C}(16)$ | 364(1) | 1550 (9) | 6 862(4) |
| C(17) | 538(1) | -388(8) | 6 143(4) |
| C(18) | $1014(2)$ | $1743(10)$ | 4 613(4) |
| C(19) | $3044(2)$ | 2842 (9) | 4 623(4) |

temperature factors, $R$ was 0.093 . At this stage, a threedimensional difference-Fourier synthesis was calculated and the positions of the 30 hydrogen atoms were found. The final five cycles of full-matrix least-squares refinement, which were carried out with anisotropic temperature factors, decreased $R$ to 0.057 for the 2084 observed reflections. In this refinement the function minimized was $\Sigma w\left(F_{\mathrm{o}}-F_{\mathrm{c}}\right)^{2}$ with unit weights; hydrogen atoms were included in the least-squares calculations with isotropic temperature factors ( $B$ fixed at $1.948 \AA^{2}$ ).
*See Notice to Authors No. 7 in J.C.S. Perkin II, 1976, Index issue.
${ }^{8}$ H. Koyama and K. Okada, 1970, SEARCHER, A Fortran program for the automatic heavy-atom analysis of organic compounds, Shionogi Research Laboratory, Osaka, Japan; see also J. Chem. Soc. (B), 1969, 940
${ }^{\circ}$ H. Koyama and K. Okada, Acta Cryst., 1970, B28, 444.

Observed and calculated structure amplitudes and anisotropic thermal parameters are listed in Supplementary Publication No. SUP 21926 ( $7 \mathrm{pp} ., 1$ microfiche).* Atomic scattering factors used in all calculations were taken from ref. 11 for non-hydrogen and from ref. 12 for hydrogen atoms. Final positional parameters, with their estimated standard deviations, are given in Tables 1 and 2.

Table 2
Fractional atomic co-ordinates ( $\times 10^{3}$ ) for the hydrogen atoms, labelled according to the atom to which they are bonded

| Atom | $x$ | $y$ | $z$ |
| :--- | ---: | ---: | :---: |
| H(la) | 336 | -109 | 465 |
| H(lb) | 339 | -155 | 581 |
| H(2) | 433 | -30 | 464 |
| H(3) | 463 | 265 | 562 |
| H(4a) | 389 | 477 | 600 |
| H(4b) | 406 | 385 | 707 |
| H(5) | 330 | 145 | 724 |
| H(6a) | 303 | 491 | 769 |
| H(6b) | 291 | 528 | 662 |
| H(7a) | 206 | 445 | 747 |
| H(7b) | 225 | 227 | 800 |
| H(8) | 200 | 351 | 573 |
| H(9) | 246 | -39 | 652 |
| H(11a) | 217 | 16 | 425 |
| H(11b) | 235 | -183 | 482 |
| H(12a) | 140 | -246 | 460 |
| H(12b) | 151 | -266 | 583 |
| H(14) | 147 | 37 | 718 |
| H(15a) | 91 | 431 | 665 |
| H(15b) | 97 | 333 | 775 |
| H(16a) | 7 | 252 | 646 |
| H(16b) | 15 | 119 | 761 |
| H(17) | 59 | -166 | 664 |
| H(18a) | 81 | 109 | 400 |
| H(18b) | 84 | 300 | 474 |
| H(18c) | 132 | 266 | 434 |
| H(19a) | 289 | 202 | 396 |
| H(19b) | 285 | 408 | 473 |
| H(19c) | 337 | 347 | 444 |
| H[O(2)] | 12 | -167 | 493 |

## DISCUSSION

The configuration of the molecule is illustrated ${ }^{13}$ in Figure 1. The steroid skeleton consists of the thiiran


Figure 1 An ORTEP drawing of the molecule, showing the atom numbering system used
ring and the basic $5 \alpha$-androstane moiety with rings A D trans-connected. Atoms $\mathrm{C}(18), \mathrm{C}(19)$, and $\mathrm{O}(2)$ are
10 W. L. Busing, K. O. Martin, and H. A. Levy, ORFLS, A Fortran crystallographic least-squares program, Oak Ridge National Laboratory, Tennessee, Report ORNL TM 305, 1962.

11 P. A. Doyle and P. S. Turner, Acta Cryst., 1968, A24, 390.
${ }^{12}$ R. F. Stewart, E. R. Davidson, and W. T. Simpson, J. Chem. Phys., 1965, 42, 3175.
${ }_{13}$ C. K. Johnson, ORTEP, A Fortran thermal-ellipsoid plot program, Oak Ridge National Laboratory, Tennessee, Report ORNL 3794, 1965.
on the $\beta$ side of the steroid skeleton, while the sulphur atom is on the $\alpha$ side.

Intramolecular bond distances and angles are given in Table 3, together with their estimated standard deviations computed from the least-squares residuals.

Table 3
Interatomic distances ( $\AA$ ) and angles $\left({ }^{\circ}\right)$, with standard deviations in parentheses

| (a) Bond distances |  |  |  |
| :---: | :---: | :---: | :---: |
| $\mathrm{S}-\mathrm{O}(1)$ | $1.503(3)$ | $\mathrm{C}(8)-\mathrm{C}(14)$ | 1.529(5) |
| $\mathrm{S}-\mathrm{C}(2)$ | 1.839(4) | $\mathrm{C}(9)-\mathrm{C}(10)$ | 1.561 (5) |
| $\mathrm{S}-\mathrm{C}(3)$ | $1.832(5)$ | $\mathrm{C}(9)-\mathrm{C}(11)$ | $1.537(6)$ |
| $\mathrm{O}(2)-\mathrm{C}(17)$ | $1.407(5)$ | $\mathrm{C}(10)-\mathrm{C}(19)$ | $1.522(6)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.530(5)$ | $\mathrm{C}(11)-\mathrm{C}(12)$ | 1.528(5) |
| $\mathrm{C}(1)-\mathrm{C}(10)$ | $1.538(6)$ | $\mathrm{C}(12)-\mathrm{C}(13)$ | 1.514(6) |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | 1.483(7) | $\mathrm{C}(13)-\mathrm{C}(14)$ | $1.537(6)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | 1.495(7) | C(13)-C(17) | 1.559 (5) |
| C(4)-C(5) | $1.538(5)$ | $\mathrm{C}(13)-\mathrm{C}(18)$ | $1.515(7)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | 1.509 (6) | $\mathrm{C}(14)-\mathrm{C}(15)$ | 1.534(6) |
| $\mathrm{C}(5)-\mathrm{C}(10)$ | $1.550(6)$ | $\mathrm{C}(15)-\mathrm{C}(16)$ | $1.527(6)$ |
| $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.518(6)$ | $\mathrm{C}(16)-\mathrm{C}(17)$ | 1.542(7) |
| $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.515(6)$ |  |  |
| $\mathrm{C}(8)-\mathrm{C}(9)$ | 1.551 (5) |  |  |
| (b) Valency angles |  |  |  |
| $\mathrm{O}(1)-\mathrm{S}-\mathrm{C}(2)$ | 110.4(2) | $\mathrm{C}(1)-\mathrm{C}(10)-\mathrm{C}(9)$ | 109.0(3) |
| $\mathrm{O}(1)-\mathrm{S}-\mathrm{C}(3)$ | 110.0(2) | $\mathrm{C}(1)-\mathrm{C}(10)-\mathrm{C}(19)$ | 108.6(3) |
| $\mathrm{C}(2)-\mathrm{S}-\mathrm{C}(3)$ | 47.6(2) | $\mathrm{C}(5)-\mathrm{C}(10)-\mathrm{C}(9)$ | 107.7(3) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(10)$ | 114.0(3) | $\mathrm{C}(5)-\mathrm{C}(10)-\mathrm{C}(19)$ | 112.0(3) |
| $\mathrm{S}-\mathrm{C}(2)-\mathrm{C}(1)$ | 117.1(3) | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(19)$ | 111.4(3) |
| $\mathrm{S}-\mathrm{C}(2)-\mathrm{C}(3)$ | 65.9(2) | $\mathrm{C}(9)-\mathrm{C}(11)-\mathrm{C}(12)$ | 114.2(3) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 119.7(3) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)$ | 111.0(3) |
| $\mathrm{S}-\mathrm{C}(3)-\mathrm{C}(2)$ | 66.4(2) | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | 108.1(3) |
| S-C(3)-C(4) | 120.7 (3) | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(17)$ | 114.5(3) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | $121.0(3)$ | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(18)$ | 111.4(3) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 114.0(4) | $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(17)$ | 98.8(3) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 110.0(3) | $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(18)$ | 113.4(4) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(10)$ | 112.2(3) | $\mathrm{C}(17)-\mathrm{C}(13)-\mathrm{C}(18)$ | 109.7(3) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(10)$ | $112.8(3)$ | $\mathrm{C}(8)-\mathrm{C}(14)-\mathrm{C}(13)$ | 113.1(3) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | $113.2(4)$ | $\mathrm{C}(8)-\mathrm{C}(14)-\mathrm{C}(15)$ | 119.8(3) |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | $111.7(3)$ | $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | 104.5(3) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | 109.9(3) | $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)$ | 103.7(4) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(14)$ | 111.4(3) | $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | 106.9(3) |
| $\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{C}(14)$ | 108.7(3) | $\mathrm{O}(2)-\mathrm{C}(17)-\mathrm{C}(13)$ | 116.3(4) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | $112.3(3)$ | $\mathrm{O}(2)-\mathrm{C}(17)-\mathrm{C}(16)$ | 108.4(3) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(11)$ | 112.3(3) | $\mathrm{C}(13)-\mathrm{C}(17)-\mathrm{C}(16)$ | 104.3(3) |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(11)$ | 113.9(3) |  |  |
| $\mathrm{C}(1)-\mathrm{C}(10)-\mathrm{C}(5)$ | 107.8(3) |  |  |
| (c) Bond distances ( $\AA$ ) associated with hydrogen positions |  |  |  |
| $\mathrm{C}(1)-\mathrm{H}(1 \mathrm{a})$ | 0.97 | $\mathrm{C}(12)-\mathrm{H}(12 \mathrm{a})$ | 1.00 |
| $\mathrm{C}(1)-\mathrm{H}(1 \mathrm{~b})$ | 0.83 | $\mathrm{C}(12)-\mathrm{H}(12 \mathrm{~b})$ | 0.97 |
| $\mathrm{C}(2)-\mathrm{H}(2)$ | 1.00 | $\mathrm{C}(14)-\mathrm{H}(14)$ | 0.97 |
| $\mathrm{C}(3)-\mathrm{H}(3)$ | 0.89 | $\mathrm{C}(15)-\mathrm{H}(15 \mathrm{a})$ | 1.04 |
| $\mathrm{C}(4)-\mathrm{H}(4 \mathrm{a})$ | 1.01 | $\mathrm{C}(15)-\mathrm{H}(15 \mathrm{~b})$ | 0.89 |
| $\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~b})$ | 0.82 | $\mathrm{C}(16)-\mathrm{H}(16 \mathrm{a})$ | 1.01 |
| $\mathrm{C}(5)-\mathrm{H}(5)$ | 0.95 | $\mathrm{C}(16)-\mathrm{H}(16 \mathrm{~b})$ | 1.04 |
| $\mathrm{C}(6)-\mathrm{H}(6 \mathrm{a})$ | 0.86 | $\mathrm{C}(17)-\mathrm{H}(17)$ | 1.00 |
| $\mathrm{C}(6)-\mathrm{H}(6 \mathrm{~b})$ | 0.89 | $\mathrm{C}(18)-\mathrm{H}(18 \mathrm{a})$ | 0.94 |
| $\mathrm{C}(7)-\mathrm{H}(7 \mathrm{a})$ | 0.81 | $\mathrm{C}(18)-\mathrm{H}(18 \mathrm{~b})$ | 0.89 |
| $\mathrm{C}(7)-\mathrm{H}(7 \mathrm{~b})$ | 1.08 | $\mathrm{C}(18)-\mathrm{H}(18 \mathrm{c})$ | 0.96 |
| $\mathrm{C}(8)-\mathrm{H}(8)$ | 1.01 | $\mathrm{C}(19)-\mathrm{H}(19 \mathrm{a})$ | 1.00 |
| $\mathrm{C}(9)-\mathrm{H}(9)$ | 0.93 | $\mathrm{C}(19)-\mathrm{H}(19 \mathrm{~b})$ | 0.89 |
| $\mathrm{C}(11)-\mathrm{H}(11 \mathrm{a})$ | 1.07 | $\mathrm{C}(19)-\mathrm{H}(19 \mathrm{c})$ | 0.87 |
| $\mathrm{C}(11)-\mathrm{H}(11 \mathrm{~b})$ | 0.84 | $\mathrm{O}(2)-\mathrm{H}[\mathrm{O}(2)]$ | 0.78 |

Mean estimated standard deviations are ca. $0.006 \AA$ and $0.3^{\circ}$. Most observed bond distances and angles are comparable to those found in other $\boldsymbol{5} \boldsymbol{\alpha}$-androstane type
${ }^{14}$ C. M. Weeks, A. Cooper, and D. A. Norton, Acta Cryst., 1971, B27, 1562.
${ }^{15}$ N. Mandel and J. Donohue, Acta Cryst., 1972, B28, 308.
${ }^{16}$ A. Chiaroni and C. Pascard-Billy, Acta Cryst., 1972, B28, 1085.
${ }^{17}$ A. Precigoux, B. Busetta, C. Courseille, and M. Hospital, Cryst. Struct. Comm., 1972, 1, 265.
steroids. ${ }^{14-21}$ Two or three differences which appear to be significant in terms of the estimated standard deviations are more likely to indicate a slight underestimation of errors.

## Table 4

Equations of planes and in square brackets displacements $(\AA)$ of the atoms from the mean planes: $X, Y$, and $Z$ are orthogonal co-ordinates $(\AA)$
Plane (I): S, C(2), C(3)

$$
-0.8264 X+0.0200 Y+0.5626 Z=-4.1399
$$

$$
[\mathrm{O}(1)-1.39, \mathrm{H}(\mathrm{C} 2)-0.90, \mathrm{H}(\mathrm{C} 3)-0.76]
$$

Plane (II): C(1)-(4)

$$
-0.1369 X+0.5126 Y-0.8475 Z=-6.7092
$$

$$
[\mathrm{C}(1)-0.00, \mathrm{C}(2) 0.01, \mathrm{C}(3)-0.01, \mathrm{C}(4) 0.00, \mathrm{C}(5)-0.29,
$$

$$
\mathrm{C}(10) 0.47]
$$

Plane (III): C(6), C(7), C(9), C(10)

$$
\begin{aligned}
& \quad-0.3710 X+0.6066 Y-0.7030 Z=-6.8168 \\
& {[\mathrm{C}(5)-0.64, \mathrm{C}(6)-0.01, \mathrm{C}(7) 0.01, \mathrm{C}(8) 0.67, \mathrm{C}(9)-0.01} \\
& \mathrm{C}(10) 0.01]
\end{aligned}
$$

Plane (IV): C(8), C(11), C(12), C(14)

$$
-0.3824 X+0.5558 Y-0.7380 Z=-6.5506
$$

$$
[\mathrm{C}(8) 0.01, \mathrm{C}(9)-0.60, \mathrm{C}(11)-0.01, \mathrm{C}(12) 0.01, \mathrm{C}(13) 0.71
$$

$$
\mathrm{C}(14)-0.01]
$$

Plane (V): C(15)-(17)

$$
-0.1836 X+0.5271 Y-0.8296 Z=-6.4596
$$

$$
[\mathrm{C}(13) 0.51, \mathrm{C}(14)-0.22]
$$



Figure 2 Projection of the molecule (i) and parallel to and (ii) in the plane of $C(8), C(10)$, and $C(13)$

The mean $\mathrm{C}-\mathrm{C}$ single-bond distance in the steroid skeleton is $1.530 \AA$. The C-S bonds ( 1.839 and $1.832 \AA$ ) in the thiiran ring are significantly longer than the similar distances ( 1.775 and $1.778 \AA$ ) found in $2 \alpha, 3 \alpha-$ epithio- $5 \alpha$-androstan-17 $\beta$-yl $p$-bromobenzoate. ${ }^{5}$ The $\mathrm{C}-\mathrm{H}$ distances range from 0.81 to $1.08 \AA$, the mean ${ }^{18}$ V. M. Coiro, E. Giglio, A. Lucano, and R. Puliti, Acta Cryst., 1973, B29, 1404.
${ }_{19}$ D. N. Peck, D. A. Langs, C. Eger, and W. L. Duax, Cryst. Struct. Comm., 1974, 3, 573.
${ }^{20}$ V. Cody, W. L. Duax, C. M. Weeks, and M. E. Wolff, Acta Cryst., 1975, B81, 292.
${ }_{21}$ D. F. Rendle and J. Trotter, J.C.S. Perkin II, 1975, 1361.
( $0.95 \AA$ ) being shorter than the internuclear separations ( $1.08 \AA$ ) obtained spectroscopically. ${ }^{22}$

The bond angles in the thiiran ring are close to their expected value. The mean interbond angle in rings $A-C$ is $111.4^{\circ}$, except for the angles at atoms $C(2)$ and $\mathrm{C}(3)$, which are slightly, but not significantly, greater than the normal tetrahedral angle of $109.4^{\circ}$. The angles in the five-membered ring D are very similar to the corresponding angles in other steroids. Maximum distortion of bond angles occurs at $\mathrm{C}(13)$ in ring D . The
steroid skeleton, calculated as the distance between $\mathrm{C}(3)$ and $\mathrm{O}(2)$, is $9.872 \AA$. The dihedral angle between the thiiran ring and the least-squares plane through atoms $\mathrm{C}(1)-(4)$ is ca. $110.7^{\circ}$. The torsion angle $\mathrm{H}(2)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{H}(3)$ is $c a .-0.1^{\circ}$.
The packing of molecules in the unit-cell is illustrated in Figure 3. The average plane of the steroid skeleton is nearly parallel to the ac plane of the cell. The two molecules are linked by the intermolecular hydrogen bond, $\mathrm{O}(1) \cdots \mathrm{H}-\mathrm{O}\left(2^{\mathrm{I}}\right)(2.775 \AA)$. All intermolecular


Figure 3 Packing of the structure viewed down the $b$ axis
angle $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(17)$ of $98.8^{\circ}$ is an indication of strain imposed at the $\mathrm{C}-\mathrm{D}$ junction.

The displacements of atoms from some least-squares mean planes are listed in Table 4. Ring a of the steroid skeleton has a distorted half-chair conformation. Atoms $C(1)$-(4) are planar and atoms $C(5)$ and $C(10)$ are displaced by -0.29 and $0.47 \AA$ to opposite sides to this plane. Rings $\bar{B}$ and c are in the normal chair form, while ring D has a conformation somewhere between a half-chair and a $\beta$ envelope.

In order to obtain detailed information on the molecular shape and the conformational features we have calculated the displacements from the planes through $\mathrm{C}(8), \mathrm{C}(10)$, and $\mathrm{C}(13)$ in the steroid skeleton (Figure 2). The geometry of molecule (1) is somewhat different from that of the previously reported $2 \alpha, 3 \alpha$-epithio- $5 \alpha$-andro-stan-17 $\beta$-yl $p$-bromobenzoate. The largest differences occur in the $\mathrm{C}-\mathrm{S}$ bond length, and are due to the presence of the S-O bond in (1). The distance ( $4.680 \AA$ ) between the $C(18)$ and $C(19)$ axial methyl groups is comparable to that found in other steroids. The width of the
distances were calculated, and the most significant contacts ( $<3.60 \AA$ ) are given in Table 5. The molecular

Table 5
Intermolecular distances ( $<3.600 \AA$ )

| $\mathrm{S} \cdots \mathrm{O}\left(2^{\text {I }}\right.$ ) | 3.501 | $\mathrm{O}(1) \cdots \mathrm{C}(18 \mathrm{III})$ | 3.508 |
| :---: | :---: | :---: | :---: |
| $\mathrm{O}(1) \cdots \mathrm{O}\left(2^{\text {I }}\right.$ ) | 2.775 | $\mathrm{C}(2) \cdots \mathrm{O}\left(2^{\text {I }}\right.$ ) | 3.542 |
| $\mathrm{O}(1) \cdots \mathrm{C}\left(\mathbf{4 I I}^{\text {) }}\right.$ | 3.432 | $\mathrm{C}(3) \cdots \mathrm{O}\left(2^{\text {III }}\right)$ | 3.378 |

Roman numerals as superscripts refer to the following equivalent positions relative to the reference molecule at $x, y, z$ :

$$
\begin{aligned}
\text { I } \frac{1}{2}+x,-\frac{1}{2}-y, 1-z & \text { III } \frac{1}{2}+x, \frac{1}{2}-y, 1-z
\end{aligned}
$$

packing is efficient with several contacts approaching the sum of the van der Waals radii; the shortest nonhydrogen contact is $\mathrm{C}(3) \cdots \mathrm{O}\left(2^{\mathrm{III}}\right) 3.378 \AA$.

We thank Dr. T. Komeno for suggesting this investigation, and S. Ishihara for the preparation of suitable crystals.
[6/1390 Received, 16th July, 1976]
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